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ORGANIC ACID TRANSPORT SYSTEM  
OF THE KIDNEY TUBULE CELL

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YALE



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EFFECTS OF ANESTHETICS ON THE ORGANIC ACID TRANSPORT SYSTEM  
OF THE KIDNEY TUBULE CELL

Submitted in partial  
fulfillment of the  
requirements for the  
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Class of 1970  
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REPORTS OF AMERICAN MEDICAL ASSOCIATION  
OF THE AMERICAN MEDICAL ASSOCIATION

















The effect of anesthetics on renal tubular transport systems has not been previously studied free of the complicating factors of changing systemic blood flow, intra-renal flow distribution, levels of circulating hormones, and so forth. Previous work in this area of anesthetic pharmacology has been indirect, and has frequently been performed with preparations in vivo, where all of these complicating problems prevail. There have been many studies which utilize the transport mechanisms, assuming without proof that they function unaltered during anesthesia and surgery. Many of the clearance techniques for study of the kidney involve tubular secretion of some substance. This is illustrated by the following table:

<u>Technique</u>	<u>Secreted Substance</u>
Effective Renal Plasma Flow (ERP <sup>2</sup> )	Para-amino-hippurate (PAH)
Excretory Urography	Diodrast, Uroselectan, Topax Neo-iopax, Skiodan
Radio-isotope Renography	<sup>131</sup> I -hippurate
Excretion Rate	Phenolsulfonphthalein (PSP) (Pitts 1963)

All of these methods depend upon normal function of the "organic-acid active transport system" of the renal proximal tubular cell, which excretes the above substances and creatinine, chlorothiazide, probenecid, and a variety of glucuronides and sulfate esters. (Pitts 1963)



This system also excretes mersalyl, an acidic mercurial diuretic. The transport mechanism may not be directly involved with the diuretic action of the drug, for non-acidic congeners, such as chlormerodrin, have identical effects on sodium excretion without being themselves similarly transported. (Cefruny and Gussin 1967)

The system exhibits competition, reversible by mass action, between the various compounds transported. This phenomenon was recognised during the war years and capitalized on to conserve penicillin when it was so scarce. (Beyer et al. 1944; Pitts 1963) These competitive interactions can influence the function of diuretics of another chemical class. "Inhibition of furosemide-induced natruresis by probenecid suggests that adequate cellular concentration (and/or luminal concentration) rather than plasma concentration is a prerequisite to the natruretic action." (Hook and Williamson 1965) The dynamics of competition may even imoinge upon the culinary artist, for excretion of the food dyes Amaranth (F.D. & C. Red No. 2), sunset yellow (F.D. & C. Yellow No. 6), and tartrazine (F.D. & C. Yellow No. 5) is through the same organic-acid system, and may be inhibited by probenecid or 4-amino-hippurate. (Despopoulos 1968)

This transport system is an active one, and is in consequence energy-dependent. (Pitts 1963) "Relatively low concentrations of azide, cyanide, and arsenite, as well as cold and anaerobiosis, block" the transport process. "All interfere with the oxidative processes which serve as sources





of energy to turn the secretory machinery. Dinitrophenol (DNP), which uncouples oxidation and phosphorylation, also "depresses" the system. (Pitts 1963) DNP itself is actively transported by this same system, and its carriage may in turn be competitively inhibited by PAH. (Bernst and Grote 1968)

Another transport mechanism for "organic bases" also exists. Tetra-ethyl-ammonium and hexamethonium, potent ganglionic blocking agents, are excreted by it. The organic base secretory mechanism exhibits competitive phenomena. (Smith et al. 1967) Also, "morphine is transported by the base transport system of the kidney" and may be partially "metabolized by the renal tubular cells." (May et al. 1967) Yet another compound of interest to anesthesiologists secreted by this route is tolazoline, an  $\alpha$ -adrenergic competitive blocker, used clinically prior to the introduction of phentolamine. (Goodman and Gilman 1965)

The two transport systems have similar mechanism, but are distinct in the compounds carried. They do not interact. Both show competitive effects within their respective chemical classes, and both appear to be in the proximal tubule. (Pitts 1963; Burg et al. 1966; Chambers and Kenton 1966) There are analogous secretory mechanisms in the liver (Despopoulos and Soenenberg 1967) and in the ciliary body of the eye. (Sugiki et al. 1961) There may be others. Careful study has shown that "the cellular transport mechanism is fundamentally similar in the pro-, meso-, and metanephric kidneys." (Jaffee 1964)



## SECRETORY MECHANISMS AND ANESTHESIA

Not long after its discovery, renal secretion was applied to the study of the pharmacology of diethyl ether. Miller and Cabot (1915) observed depression of phenolsulphone-phthalein excretion, "Excretion Rate," with ether anesthesia and operation. The depression was increased by shock. Many subsequent studies have either used some measure dependent on renal tubular function or been directed at the secretory function itself.

Coller et al. (1943) used Diodrast clearance ( $C_{\text{Diodrast}}$ ) in the study of effects of ether and cyclopropane on renal function in operated man with morphine and scopolamine as premedication, and found little change of  $C_{\text{Diodrast}}$  with anesthesia, in the absence of major shifts in blood volume. Craig et al. (1945) found that their "deep" ether and cyclopropane anesthesia in dogs produced depression of urine flow, both glomerular filtration and renal blood flow (as measured by clearance techniques), and the maximal tubular transport of glucose, while "light" anesthesia had little effect. Burnett et al. (1949) reported a study in man, using mannitol clearance for glomerular filtration rate (GFR), and PAH for ERPF, comparing ether and cyclopropane before operation in unpremedicated patients in the first to second plane of the third stage of anesthesia, by clinical criteria. They noted substantial falls in ERPF, averaging 39% with ether and 52% with cyclopropane, in the absence of hypotensive episodes.





More recently, Habib et al. (1951) published observations on meperidine, cyclopropane, ether, and thiopental in man. They found substantial vasoconstriction of the renal vessels, as determined by clearance studies, and found that operation added little additional effect. Glauser and Selkurt (1952) studied barbiturates in dogs, finding that both pentobarbital and barbital reduced plasma flow in the kidney as measured by clearance techniques, 18% and 16% respectively. Miles and DeWardener (1952) reported that both ether and cyclopropane produced vasoconstriction in kidneys, using flowmeters rather than clearance methods. Miles et al. (1952) reported on the interaction of pentamethonium bromide with ether and cyclopropane. The depression of renal blood flow (measured by clearance techniques) by the two inhalants was not increased by the addition of the blocking agent. Hill et al. (1958) examined the renal effects of meperidine, phenobarbital, and chloral hydrate, and not only used low-level PAH excretion for ERPF, but also studied the maximal transport capabilities of the tubules ( $T_M^{PAH}$ ), showing awareness that there might be complex dynamics in renal tubular function.

There has been further expansion of the work in the past decade, and an increasing utilization of more sophisticated methods. Mazze et al. (1963) presented data on halothane, and in man, found with standard PAH clearance measurements that there was a dose-response pattern.  $C_{PAH}$  was 60% of control



for 0.5-1.0% halothane but was further decreased to 48% of control values for 1.2-3.0%. Their patients were premedicated, with morphine and scopolamine, and were operated upon. In a review of renal effects of drugs used in anesthesia, with presentation of some new data, Papper and Papper (1964) suggested that it may be essential to measure renal vein PAH in order to use the standard clearance techniques with reliability, as there may be changes in extraction due to drug effects. Miller et al. (1965) and Miller et al. (1966) pointed out that the original data on halothane had included the pathologic finding in dogs of proximal tubular dilatation (without necrosis and without histochemical alteration in alkaline phosphatase or non-specific esterase, Raventós 1956). They studied  $T_M$ PAH in detail, in unpremedicated patients without use of barbiturates or relaxants. The first study suggested some depression of tubular function by halothane, but the second study, taking account of changes in blood pressure, showed only changes proportional to the levels of hypotension induced, both by halothane and by halothane-nitrous oxide mixture.

Deutsch et al. (1966) also reported on the effects of halothane. They used ethanol infusions to overcome the antidiuretic state often associated with anesthesia, so that urine flow would be adequate for reliable clearance studies. Their patients were not premedicated and were studied prior to the initiation of operative manipulation. In addition,



renal vein PAH was determined in several, to assess the propriety of using PAH clearance for ERPF. With 1.5% halothane, they observed 38% decrease in renal blood flow. Deutsch et al. (1967) in similar experiments found that cyclopropane 19% (in end-expiratory air) produced a 42% depression in renal blood flow. Renal vein PAH was not measured in this study. Kennedy et al. (1969) reported that peridurals, using local anesthetics, reduced GFR in proportion to the accompanying decrease in blood pressure, but reduced ERPF even more. The effect was increased by the addition of epinephrine to the injection. They used  $^{57}\text{Co}$  cyanocobalamin for GFR, following loading dose of cold B12, and  $^{125}\text{I}$ -Hippuran for measuring ERPF.

As may be readily concluded, only the study by Deutsch et al. (1966) made the critical test of measuring renal vein PAH so that extraction could be ascertained. The data of Miller et al. from 1965 on transport maxima must be rejected for failure to control perfusion pressure, but the 1966 experiments by Miller et al. were suitably controlled. Most of the other studies have one or more practical or theoretical objections which invalidate them as detailed studies of renal tubular secretory function, suggesting that perhaps a more direct approach is necessary. This may be accomplished through examining the function of isolated tubules directly, thereby eliminating any question of indirect effects of the anesthetics by change of blood pressure, by change of intra-renal blood flow, by release of catecholamines, or by any other means.





## ISOLATED TUBULE STUDIES

In the course of some basic research on tissue culture of chick kidneys, Chambers and Kempton (1933) observed the concentration of phenol red by cystic elements formed by the culture. They reported detailed studies of the phenomenon, stating that only cysts formed from proximal tubule of either the mesonephric  $4\frac{1}{2}$  day old embryo or the metanephric newly hatched chick would concentrate the dye in their lumina, and that the effect could be blocked by reducing the temperature to  $3-6^{\circ}\text{C}$ , suggesting some metabolic involvement.

Forster (1948) reported work with several types of kidneys, in thin slices, and also observed that fish kidneys could readily be separated into individual tubules, as they have less interstitial tissue than higher vertebrate kidneys. At that time, he described direct microscopic observation of these isolated tubules, analogous to the chick work, and gave some basic data on inhibition of phenol red transport by a variety of compounds. As further elucidated by Forster and Teggart (1950), the microscopic technique involved placing the isolated tubules into a bath of fish Ringers with a 1 mg% solution of phenol red. They graded intra-luminal concentration of the dye in the tubules and found that the time to first discernible concentration was 2-5 minutes in controls. The time to peak concentration was typically 30-60 minutes. Some tubules failed to function, and others performed poorly, presumably due to the trauma of disrupting the kidneys. The best functioning



units were used for grading. Many of the inhibiting chemicals mentioned earlier were first studied in this paper, including cyanide, azide, arsenite, phlorhizin, iodoacetate, fluoride, DNP, and mercury salts, as well as cold and anaerobiosis. They also described many of the competitive inhibitors, including PAH, penicillin, carinamide, and some antidiuretic derivatives of cinchoninic acid. Peck et al. (1952) added to the data on the effects of various ions, using the same experimental system, and found that the transport was apparently a two step process. Transport from the bath to the cell requires  $K^+$  in the medium, and may be inhibited competitively with PAH, whereas the further transport from the cell to the lumen of the tubule requires the presence of  $Ca^{++}$  in the medium, and is not subject to competitive inhibition.

While studying the transport of a large series of closely related phenolsulfonphthalein dyes, Forster et al. (1954) observed in the same experimental arrangement that bromocresol green, the dye least efficiently transported, proved to be the most potent inhibitor of the others. They also noted that chlorphenol red "resulted in more uniform preparations than were obtained with phenol red. The bluish-red color of the former is more readily perceived and the stability of the color within the pH encountered in these preparations is a distinct advantage." Consequently, most subsequent work has used chlorphenol red in this experimental format.



The method was applied to the study of the two step nature of the transport by Forster and Copenhaver (1956) with adaptation to rabbit cortex slices. Forster and Hong (1958) and Hong and Forster (1958) again applied it, using flounder tubules. They added a "runout" technique, placing saturated tubules in medium free of dye, and observing subsequent changes. Hong and Forster (1959), again with flounders and the added runout technique, found that cell to lumen movement under these conditions could be influenced by  $\text{Ca}^{++}$ , phlorhizin, and competitive inhibitors. Forster and Goldstein (1961) reported a significant correlation of the activity of this transport system, as determined by  $\text{T-PAH}_M$ , with renal succinoxidase activity, this enzyme being part of the Krebs cycle for aerobic metabolism. They studied a wide range of animals, including goosefish (Lophius sp.), frog, dog, cat, and rat. Burg et al. (1966) showed that the single tubule could be perfused under suitable microscopic controls, and demonstrated active PAH transport in proximal tubules of rabbits. Winter (1966) took the standard fish method to a high level of sophistication by arranging an adaptation to microspectrophotometry, and procured objective confirmation of earlier visual results.

Despite the reproducibility of the technique of using isolated fish tubules, certain anatomic questions remained. Concentration in the lumen implied that both ends of the open cylinder of an isolated tubule must spontaneously seal. The necessary proof of this, as well as data on the length of time an isolated tubule could remain morphologically unchanged, was



provided by Bulger and Trump(1965) and Trump and Bulger (1967). Their early study involved primarily the normal state of the sole, but their latter report showed many details of the transport phenomena. They found that fine structure, by light and electron microscopy, was maintained at least 6 hours in oxygenated balanced salt solution at 0-4°C, as was dye transport when the specimen was temporarily returned to 25°C. There was good correlation of capacity for dye transport with the maintenance of ultrastructure. Tubules showing damage showed no transport. Autophagocytosis was the principle injury reaction.

Several detailed reviews of renal transport mechanisms are of interest, summarizing the laboratory data and suggesting theoretical bases for understanding the phenomena, including Taggart (1958), Weiner and Nudge (1964), and Forster (1967). They describe methodology as well as results, primarily in terms of a carrier-mediated transport model.





## EXPERIMENTAL METHODS

Using the methods of Forster and Taggart (1950) with chlorphenol red in lieu of phenol red (per Forster et al. 1954), adapted to the kidney tubules of goldfish (Carrassius auratus, L.), I studied the effects of thiopental sodium, nitrous oxide, halothane, and methoxyflurane on the "organic acid" active transport system in isolated tubules in vitro.

The goldfish were purebred Shobunkin strain, obtained from the Stoutland Hatcheries, Stoutland, Missouri, as fingerlings. They were maintained in large tanks with aeration and filtration, and fed "Tetra Min" brand prepared large flake staple food, three weeks, alternating with fresh or fresh-frozen Tubifex worms, one week, until they were about one year of age. The tank water was always aged, de-chlorinated, de-fluorinated, buffered with "5-in-1 Water Conditioner" (by Aqua-Biotics, Inc.), and brought to 17-19°C before the fish were exposed to it. Feeding was at 9:00 A.M. daily and the tank room was in darkness from 5:00 P.M. until 8:30 A.M. daily. The laboratory itself was held at 18-20°C except for one day toward the end of the experiments when a power shut-down eliminated the air-conditioning and the temperature reached 22° in the tanks and 24° C in the laboratory.

The goldfish Ringer's solution had the following salts:

<u>Compound</u>	<u>mMole/liter</u>
NaCl	100.0
KCl	2.5
CaCl	1.5
MgCl <sub>2</sub>	1.0
NaH <sub>2</sub> PO <sub>4</sub>	0.5
NaHCO <sub>3</sub>	10.0



After removal, a pair of kidneys was placed in this solution and teased apart into separate tubules or clumps of tubules. These were stored in the same solution at 4°C in shallow dishes. The experimental medium was Ringer's with chlorphenol red added to a concentration of  $10^{-5}$  M plus the appropriate test substance dissolved in or perfusing the medium.

Studies were performed with the specimens in split Petri dishes (Falcon Plastics #1003 "Two Compartment Petri Dish" cut down to one-half its initial height to accommodate the short objective distance of the 4X objective of a standard Nikon SBR microscope), either with the top removed, for the thiopental studies, or with the top on and cemented along the midline divider and edges with wedge-shaped cut-outs on either side for perfusing gas entry and exit. One side of the split Petri dish was always used for simultaneous control specimens. All studies were conducted under a ventilating hood at the measured room temperature, and the sub-stage lamp of the microscope was only turned on at the time of observations. A brief trial of controls confirmed the 6 hour or more true functional viability previously demonstrated by Trump and Bulger, and indeed one of these still had functioning tubules more than 24 hours after sacrifice, although the specimen had a very large amount of autolysis at that time microscopically. As U.S.P. Sodium Thiopental for Injection is highly alkaline, experiments run with it were done in the presence of 0.01 M secondary-tertiary sodium phosphate buffer mix, pH 7.4,



which itself was tested for effect. Experiments with nitrous oxide were performed with nitrogen as a control gas, both in oxygen. Halothane and methoxyflurane were vaporized with oxygen, and hence were tested against oxygen controls. Both the time to first discernible concentration and the time of peak concentration was obtained on each trial.

When using the gas or the vapors, the apparatus was set up an hour early and perfused to allow equilibration with the medium and saturation of the connecting tubing. Halothane was vaporized with a Fluotec Mark II and methoxyflurane was vaporized with a Foregger Copper Kettle in a 20°C water bath. Control of gas flows was with conventional rotameter-type gas flow meters and standard anesthesia equipment. Flows were kept high enough for reliable vaporization, with a bleed point added to the circuit just before the bubbler to vent excess gases to the hood.

Data for each series was compared to its simultaneously conducted control series, and means and sample standard deviations computed for each. The two-sample t test was the index of significance for difference between the means. (Freund 1962) Standard tables of t values came from the Handbook of Chemistry and Physics. (Weast et al. 1964)



# EXPERIMENTAL RESULTS

TEST CONDITIONS	TIME TO FIRST DETECTABLE CONCENTRATION			
	Control+	Test	n	P*
Buffer	7.1±1.27 min.	6.8±1.64 min.	9	NS
Buffer + 20 mg% Thiopental Na	8.4±1.99	22.6±3.51	7	S
Buffer + 10 mg% Thiopental Na	6.5±0.84	9.5±1.38	6	S
80% Nitrous Oxide	9.1±1.56	13.7±3.50	12	S
1.2% Methoxyflurane in O <sub>2</sub>	5.9±1.46	8.4±2.32	8	NS
2.0% Halothane in O <sub>2</sub>	6.3±1.75	7.0±1.79	6	NS
(Derived values comparing 20 mg% Thiopental Na with 10%)	20 mg% 22.6±3.51	10 mg% 9.5±1.38	7/6	S

\*S = significant,  $P \leq 0.05$ ; NS = not significant,  $P > 0.05$   
+ Mean ± Sample Standard Deviation





# EXPERIMENTAL RESULTS, CONTINUED

TEST CONDITIONS	TIME TO MAXIMUM CONCENTRATION			P*
	Control+	Test	n	
Buffer	27.9±4.36 min.	32.6±3.80 min.	9	NS
Buffer + 20 mg% Thiopental Na	35.3±3.16	59.1±5.48	7	S
Buffer + 10 mg% Thiopental Na	39.2±1.25	59.8±8.87	6	S
80% Nitrous Oxide	37.1±3.46	43.7±8.45	12	NS
1.2% Methoxyflurane in O <sub>2</sub>	37.2±2.03	54.9±5.63	8	S
2.0% Halothane in O <sub>2</sub>	39.2±2.68	44.1±6.40	6	NS
(Derived values comparing 20 mg% Thiopental Na with 10%)	20 mg% 59.1±5.48	10 mg% 59.8±8.87	7/6	NS

\*S = significant, P ≤ 0.05; NS = not significant, P > 0.05  
+ Mean ± Sample Standard Deviation



## DISCUSSION

The agents chosen for study have wide use in modern practice. Each has been examined in concentrations met in its clinical use. Although ordinarily two-thirds of thiopental in the bloodstream is bound to plasma proteins (Goldb and Smith 1954), the free arterial concentration during rapid induction may momentarily exceed 20 mg%. (Price et al. 1960) Comparison of 20 mg% to 10 mg% suggests that there may be a dose-response relationship at least in terms of the initial effects of the drug upon renal tubular function. Nitrous oxide was studied at the high end of the clinical range of concentration, while halothane and methoxyflurane were tested at more clinical levels. Methoxyflurane at 1.2% was about 5.2 MAC (minimal alveolar concentration) while halothane at 2.0% was around 2.3 MAC. Perhaps the more striking effects of methoxyflurane compared to control was due to this more than two-fold difference in clinical potency used. Nitrous oxide was at less than half its MAC, as no pressure facilities were available. (see Eger et al. 1965 for MAC values and a discussion of their significance)

Clearly, all agents studied, with the possible exception of halothane in the concentration tested, have significant effects on renal tubular transport processes, as measured by this method. It is noteworthy that the only in vivo study with renal vein PAH measured as a test of the method was with halothane



(Deutsch et al. 1966). Therefore, most prior studies of renal blood flow, measured by the traditional clearance techniques, must be considered as possibly not valid when used in anesthetized man or experimental animal. That the drugs should have such direct effects on processes at a cellular level should not be surprising, as other actions of a similar direct nature have been previously demonstrated. (see for example Andersen 1966 for a study of the effects of anesthetics upon cellular ion transport mechanisms, and Van Dyke et al. 1964 on the cellular metabolism of the drugs themselves) Perhaps some of the debate concerning the renal effects of methoxyflurane (North and Stephen 1965, also Crandell et al. 1966) may be related to the cellular depressant effects, if indeed there is true toxicity. From the introductory discussion presented before, it is clear that the ramifications may be widespread in medicine.



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